

Intramolecular Cyclization Reaction of 1,1'-Bis-(α -hydroxyisopropyl)ferrocene

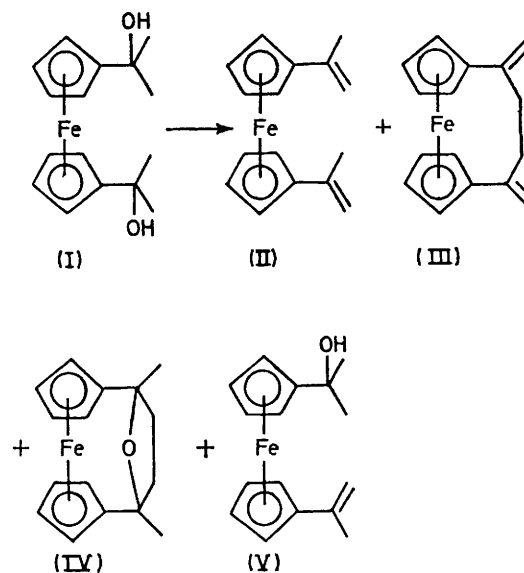
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Summary Treatment of 1,1'-bis-(α -hydroxyisopropyl)ferrocene (I) with acid gave the novel [4]ferrocenophanes (III) and (IV) together with the vinyl derivatives (II) and (V); reduction of (III) and (IV) afforded *cis*- and *trans*-6,9-dimethyl[4]ferrocenophane, which was also synthesized from (VII) by an alternative route.

THE reaction of ferrocene derivatives through stable carbonium ions, carbenes, and radicals has been studied recently.¹ We reported² that treatment of 1,1'-bis-(α -hydroxyalkyl)ferrocenes with acid initially formed stable α -hydroxyalkyl- α' -ferrocenylcarbonium ions as intermediates and then gave 7-oxa[3]ferrocenophanes by intramolecular cyclization. On the other hand, Pittman³ reported that 1,1'-bis-(α -hydroxyisopropyl)ferrocene (I) in magic acid (FSO₃H-SbF₅) had the ¹H n.m.r. signals of a dication. The product expected from the intermediate is 1,1'-bis-(α -methylvinyl)ferrocene (II).

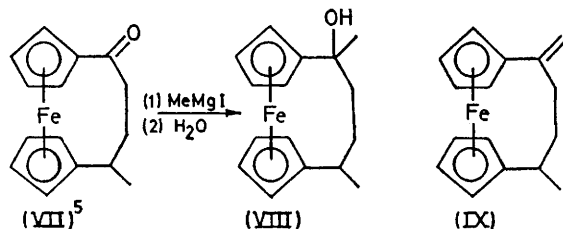
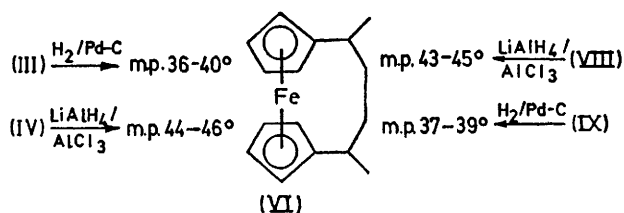
However, we found that treatment of (I) with 3-6N-HCl in benzene gave the novel [4]ferrocenophanes (III) and (IV) together with the vinyl derivatives (II) and (V). Typical results are given in the Table. Ascorbic acid and O₂ gas were added to make the conditions either oxidising or reducing.



The reaction products were separated by column chromatography into four compounds; red plates, m.p. 59–61° (II, lit.⁴ m.p. 58–59°), yellow needles, m.p. 123–124° (III), orange yellow needles, m.p. 137–138° (IV), and a red oil (V). The oil was shown to be (V) from its ¹H n.m.r. and i.r. spectra [δ 1.49(s, Me), 2.06(m, :CMe), 4.89 and 5.20 (complex AB system, :CH₂), ν_{OH} 3550 and 3450 cm⁻¹, $\nu_{C=C}$ 1613 cm⁻¹]. Compound (III) showed $\nu_{C=C}$ bands (1613 and 1628 cm⁻¹), and methylene [δ 2.55(4H, s)] and terminal methylene signals [δ 4.87 and 5.03 (4H, AB system)].

TABLE

Conditions	Reaction time (t/min)	Products (%)			
		(II)	(III)	(IV)	(V)
3N-HCl	30	10	—	15	—
4N-HCl	5	34	—	6	32
6N-HCl	30	—	6	16	—
3N-HCl+ascorbic acid ..	30	31	—	—	—
6N-HCl + O ₂	7	—	3	3	—



Compound (IV) had a C—O—C [$\nu_{\text{C-O-C}}$ 1020—1110 cm^{-1} (four peaks)], two methyl [δ 1.62(s)], and two methylene groups [δ 1.75—2.45 (AA'BB' system)]. The high-resolution mass spectrum of (IV) gave $M^+ = 282.0706$ ($\text{C}_{16}\text{H}_{18}\text{OFe}$: 282.0706). Compounds (III) and (IV) were both converted into 6,9-dimethyl[4]ferrocenophane (VI) on reduction with $\text{H}_2/\text{Pd-C}$ and $\text{LiAlH}_4\text{-AlCl}_3$, respectively. Compound (VI) was also synthesized by an alternative route: Grignard reaction of (VII)⁵ with MeMgI gave two compounds, (VIII) (oil) and (IX) (m.p. 73.5—75°), which were reduced with $\text{LiAlH}_4\text{-AlCl}_3$ and $\text{H}_2/\text{Pd-C}$, respectively, to give (VI). The reduction product of (IV) [δ 1.04(6H, d, Me), 2.33(2H, m, CH)] is assigned the *cis* configuration from the way in which it reacted, while the products from (III), (VIII), and (IX) are considered to be mixtures of *cis* and *trans* isomers from their ^1H n.m.r. spectra [δ 1.04 (d, Me) and 1.11(d, Me), 2.33(m, CH) and 2.57(m, CH)].

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¹ A. F. Forrester, S. P. Hepburn, R. S. Dunlop, and H. H. Mills, *Chem. Comm.*, 1969, 698; T. D. Turbitt and W. E. Watts, *ibid.*, 1971, 631; J. A. Connor and J. P. Lloyd, *J.C.S. Dalton*, 1972, 1470; A. Eisenstadt and M. Cais, *Chem. Comm.*, 1967, 216.

² K. Yamakawa and M. Hisatome, *J. Organometallic Chem.*, in the press.

³ C. U. Pittman, *Tetrahedron Letters*, 1967, 3619.

⁴ G. R. Knox and P. L. Pauson, *J. Chem. Soc.*, 1961, 4610.

⁵ W. M. Horspool and R. G. Sutherland, *Chem. Comm.*, 1967, 240.